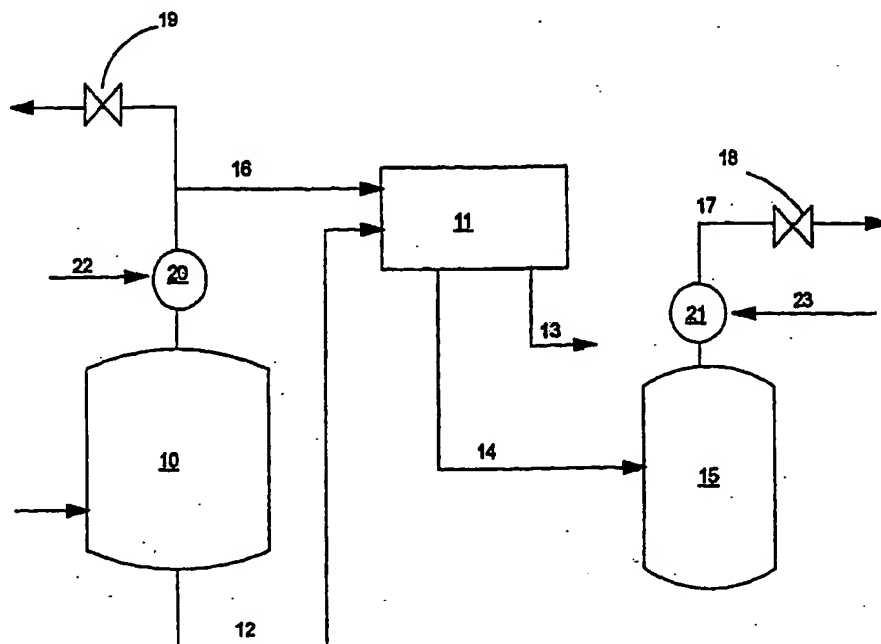


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## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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(54) Title: IMPROVED PROCESS FOR SEPARATING PURE TEREPHTHALIC ACID



(57) Abstract

An improved process for separating pure terephthalic acid crystals from a slurry in a centrifuge whereby steam is introduced as a carrier fluid into the centrifuge continuously and simultaneously with the slurry.

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TITLEIMPROVED PROCESS FOR SEPARATING PURE TEREPHTHALIC ACID

This application claims benefit of GB Provisional Application No. 9717251.4, filed August 15, 1997.

5

BACKGROUND OF THE INVENTION

The present invention relates to an improved process for separating pure terephthalic acid crystals from a slurry which contains such crystals along with dissolved impurities using a centrifuge, and, more particularly, to a process improvement whereby a carrier fluid is introduced into the centrifuge continuously and simultaneously with the slurry with the result that a higher proportion of impurities are removed from the pure terephthalic acid crystals than could have been achieved otherwise.

The production of pure terephthalic acid (PTA) involves oxidising para-xylene to form crude terephthalic acid (TA) which contains a number of impurities, particularly 4-carboxybenzaldehyde (4-CBA), and then purifying the crude TA. Crude TA crystals are dissolved in water at elevated pressure and temperature, and the solution is subjected to hydrogenation in the presence of a Group VIII noble metal hydrogenation catalyst. The purified acid is recovered by crystallizing the acid from the hydrogen treated aqueous solution, i.e., the pure plant mother liquor (PPML). The principal impurities, which are p-toluic acid derived from the compound 4-carboxybenzaldehyde and unidentified color bodies, along with some other organic components, such as benzoic acid and residual terephthalic acid, remain dissolved in the aqueous solution. The pure TA crystals are then separated from the pure plant mother liquor in one or more centrifuges, usually at elevated pressure. The present invention results in the isolation of pure TA crystals having lower levels of impurities, particularly p-toluic acid, than could have

been isolated otherwise using high pressure centrifuges.

#### SUMMARY OF THE INVENTION

The present invention is an improved process for separating pure terephthalic acid crystals from a slurry which contains such crystals along with dissolved impurities using a centrifuge. The improvement comprises introducing steam, optionally in the presence of an inert gas, such as nitrogen, as a carrier fluid into the centrifuge continuously and simultaneously with the slurry with the result that a higher proportion of the impurities, particularly p-toluic acid, are removed from the pure terephthalic acid crystals than could have been achieved otherwise. The centrifuges contemplated for use according to the invention typically include means for introducing the slurry into the centrifuge, along with a liquid and a solids output means. A difference in pressure is typically maintained between the liquid output means and the centrifuge and the solids output means of about 1-10 kPa. The separation of solid pure acid crystals from pure plant mother liquor is achieved according to the invention by introducing steam, optionally in the presence of an inert gas, such as, for example, nitrogen, into the centrifuge simultaneously and continuously with the slurry. The carrier fluid moves through the centrifuge and exits therefrom with separated pure plant mother liquor via the liquid output means. The recovered PTA crystals have an unexpectedly low concentration of p-toluic acid than PTA crystals produced without the use of a carrier fluid.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 is a simplified schematic diagram of a separation process according to the invention.

#### DETAILED DESCRIPTION

In the purification stage of typical commercial processes for producing pure terephthalic acid, i.e.,

PTA,, crude terephthalic acid crystals are dissolved in water at elevated pressure and temperature, and the solution is subjected to hydrogenation in the presence of a Group VIII noble metal hydrogenation catalyst.

5 The purified acid is recovered by crystallizing the acid from the hydrogen treated aqueous solution and then separating the pure crystals from the solution using one or more high pressure centrifuges. The separation process can be conducted batch-wise,  
10 although the process is usually carried out on a continuous basis. The principal impurities which are present in the hydrogen treated aqueous solution are p-toluic acid derived from the compound 4-carboxybenzaldehyde and various color bodies, along  
15 with some other organic components, such as benzoic acid and residual terephthalic acid. These impurities for the most part remain dissolved in the aqueous solution during the separation process. The aqueous solution which remains after separation of the pure TA  
20 crystals is referred to as "pure plant mother liquor", i.e., PPML.

The invention resides in the discovery that pure TA crystals can be separated from a pure plant mother liquor with a substantially lower concentration of p-  
25 toluic acid than could be achieved otherwise by introducing steam, optionally in the presence on an inert gas, such as nitrogen, as a carrier fluid into the centrifuge continuously and simultaneously with the slurry. Any carrier fluid which is inert with respect  
30 to the pure TA and PPML is contemplated for use in carrying out the invention.

In practice, the slurry of pure TA crystals passes into the centrifuge from a crystallizer, in which the temperature and pressure of the incoming  
35 slurry has been reduced thereby producing steam. This, in turn, can be a source of steam for practicing the invention. Residual solids which may be present in the steam as it is vented from the crystallizer can be

removed before it is introduced into the centrifuge. A water spray in the crystalliser steam vent line is suitable for this purpose.

The flow of steam, i.e., carrier fluid, through the centrifuge can be adjusted by valves.

In practice, the pressure in the liquid output means from the centrifuge is less than the pressure in the centrifuge, and this pressure differential provides a convenient method for inducing a flow of the carrier fluid through the centrifuge and facilitates its removal via the liquid output means such that said carrier fluid flows into and through the centrifuge continuously and simultaneously with the pure TA slurry. The flow of carrier fluid through the centrifuge and into the liquid output means is responsive to the reduction of pressure in the liquid output means relative to the pressure of the centrifuge.

In practice, a water spray can be directed into the liquid output means, e.g. into the vessel for containing the liquid or into the vapor line which exits such a vessel which, in turn, acts to reduce foaming in the liquid output means and thereby reduce carry over of any solids into the vapor lines.

An embodiment of the process of the invention will now be described with reference to Fig. 1, which is a simplified schematic diagram of the relevant portion of a process for purifying crude terephthalic acid.

The purification of crude terephthalic acid to produce PTA is carried out largely to remove 4-carboxy benzaldehyde (4-CBA). 4-CBA is an undesirable impurity in a PTA product because it can detrimentally affect the molecular weight of the polyester, i.e., poly(ethylene terephthalate) made therefrom. A typical specification for commercially produced PTA specifies a 4-CBA concentration of less than 20 ppm. In a typical process for removing 4-CBA, the 4-CBA is catalytically hydrogenated to p-toluic acid, which is then removed by

dissolution in water, from which solution PTA is crystallized out.

The crystallization process is multi-stage and operates at elevated temperatures and pressure. The  
5 final crystallizer is typically operated at a temperature in the range of from 140° to 155°C and at a pressure in the range of from 400 to 500 kPa, although higher and lower temperatures and pressures can be used with satisfactory results.

10 Referring now to the drawing, final crystallizer 10 produces a slurry of PTA crystals in the pure plant mother liquor. The slurry is pumped via line 12 to centrifuge 11. In practice PTA crystals are separated  
15 as a solid and recovered from the slurry in one or more centrifuges 11. Centrifuge 11 has a discharge chute for continuously removing solids via line 13 and a liquid runoff line 14 which runs to mother liquor storage tank 15, collectively "liquid output means".  
20 Spray 21 sprays water into the mother liquor storage tank 15 which reduces foaming and, in turn, improves the flow of mother liquor from the centrifuge. Solid PTA, recovered as a "cake" from the centrifuge, is re-slurried with water and re-centrifuged and then dried to form a final PTA crystalline product. The pressure  
25 in mother liquor storage tank 15 and line 14 is low relative to the pressure in centrifuge 11. Steam flows continuously from the crystallizer exit via transfer line 16 into centrifuge 11 and exits through line 14 with pure plant mother liquor. As shown, steam is  
30 generated in crystallizer 10 during operation by reducing the temperature and pressure of the incoming aqueous slurry of pure terephthalic acid. Hot water spray 20 is present as shown to remove residual solids to the extent they may be present in the vapor line  
35 being carried over from the crystallizer. The steam is returned to a steam venting system via line 17. The flow of steam from crystallizer 10 can be controlled as

shown by valves 18 & 19. Valve 19 controls the flow of steam to vent header.

The advantage of using a carrier fluid such as steam according to the invention is compared to a conventional process and illustrated in the following examples.

#### Example

A terephthalic acid purification process was operated using a crude terephthalic acid feed obtained from the oxidation of p-xylene. Crude terephthalic acid (CTA) was purified by dissolving it in water to produce an aqueous solution which was then hydrogenated in the presence of a hydrogenation catalyst whereby 4-CBA was reduced to p-toluic acid. The hydrogenated CTA was then passed through a series of crystallizers in which pure terephthalic acid crystallized out of solution leaving the p-toluic acid and other impurities in solution. The 4-CBA contained in the crude TA feed was measured at different points in time and the p-toluic acid content of the purified TA was also measured. Since the process involved conversion of 4-CBA to p-toluic acid, as the 4-CBA content of the feed increased, the amount of dissolved p-toluic acid which must be removed in the centrifuge also increased.

Table 1 shows typical levels of 4-CBA in the crude CTA feed and the p-toluic acid levels in the resulting PTA product using a centrifuge with and without introduction of a steam carrier fluid. The mother liquor output tank, i.e., "liquid output means" was maintained at a reduced pressure relative to the centrifuge. All of the PTA product contained less than 150 ppm of p-toluic acid. Before introducing steam as carrier fluid, low levels of p-toluic acid were achieved by reducing the amount of 4-CBA produced in the oxidation stage of the process. However, in doing so, some of the p-xylene feedstock and acetic acid solvent was consumed in undesirable side reactions,

which, in turn, can substantially increase the cost of operating the process on a commercial scale.

Table 1

|                                       | Before steam |          |      | After steam |          |      |
|---------------------------------------|--------------|----------|------|-------------|----------|------|
|                                       | 0.2<br>0     | 0.1<br>7 | 0.18 | 0.2<br>2    | 0.2<br>1 | 0.23 |
| 4-CBA in feed (%w/w)                  |              |          |      |             |          |      |
| p-toluic acid in PTA<br>product (ppm) | 145          | 130      | 140  | 125         | 107      | 117  |

5

The p-toluic acid levels in the PTA product after introducing steam into the centrifuge as a carrier fluid according to the invention are significantly lower, although the levels of 4-CBA in the feed are generally higher than runs measured without the carrier fluid. Thus, separation of p-toluic acid has been improved using a carrier fluid, e.g., steam, according to the invention.

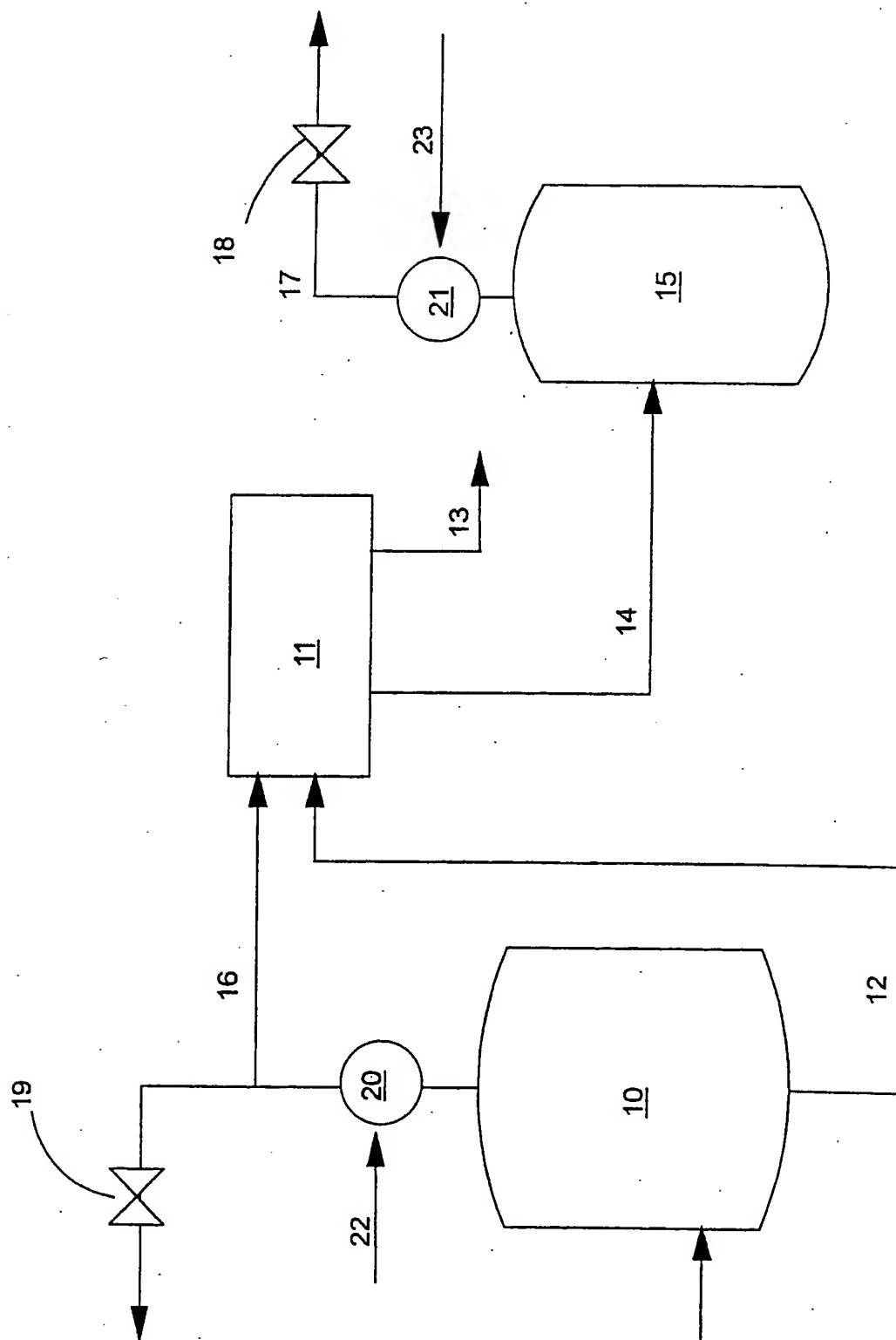
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WHAT IS CLAIMED IS:

1. In a process for purifying crude terephthalic acid comprising the steps of:
  - 5 a) dissolving said crude terephthalic acid in water at elevated temperature and pressure,
  - b) passing the resulting solution with hydrogen over a hydrogenation catalyst whereby 4-CBA is converted to p-toluic acid,
  - 10 c) reducing the temperature of the hydrogenated solution whereby pure terephthalic acid crystals crystallize out of solution to form a slurry, and
  - d) separating the pure terephthalic acid crystals from the slurry using centrifugal force in a
  - 15 centrifuge, the improvement comprising
  - e) introducing a carrier fluid into the centrifuge continuously and simultaneously with the slurry.
2. The process of Claim 1 in which the carrier fluid is steam.
- 20 3. An improved process for separating pure terephthalic acid crystals from a slurry which contains such crystals along with dissolved impurities using centrifugal force in a centrifuge wherein the improvement comprises introducing steam as a carrier
- 25 fluid into the centrifuge continuously and simultaneously with the slurry.
4. The process of claim 2 in which said steam is accompanied by an additional inert gas.
5. The process of Claim 3 in which the inert gas is
- 30 nitrogen.

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# INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 98/16691

**A. CLASSIFICATION OF SUBJECT MATTER**  
IPC 6 C07C51/43 C07C63/46

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C07C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

| Category * | Citation of document, with indication, where appropriate, of the relevant passages   | Relevant to claim No. |
|------------|--|-----------------------|
| X          | WO 93 24440 A (IMPERIAL CHEMICAL INDUSTRIES PLC) 9 December 1993<br>see page 8, line 21 - page 9, line 24<br>see page 10, line 1 - line 18<br>see page 11, line 23 - page 12, line 7<br>see page 20, line 23 - page 21, line 7<br>see page 24 - page 28; claims<br>see figure 5<br>----- | 1-4                   |

☐ Further documents are listed in the continuation of box C.

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Date of the actual completion of the international search

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Information on patent family members

International Application No

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|---|---------------------|----------------------------|---------------------|
| WO 9324440 A                              | 09-12-1993          | AU 681078 B                | 21-08-1997          |
|   |                     | AU 4081593 A               | 30-12-1993          |
|   |                     | BR 9306434 A               | 15-09-1998          |
|   |                     | CA 2136621 A               | 09-12-1993          |
|   |                     | CN 1085889 A               | 27-04-1994          |
|   |                     | EP 0642490 A               | 15-03-1995          |
|   |                     | JP 7507291 T               | 10-08-1995          |
|   |                     | MX 9303173 A               | 29-07-1994          |
|   |                     | SG 52279 A                 | 28-09-1998          |
|   |                     | US 5583254 A               | 10-12-1996          |
|   |                     | US 5698734 A               | 16-12-1997          |

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